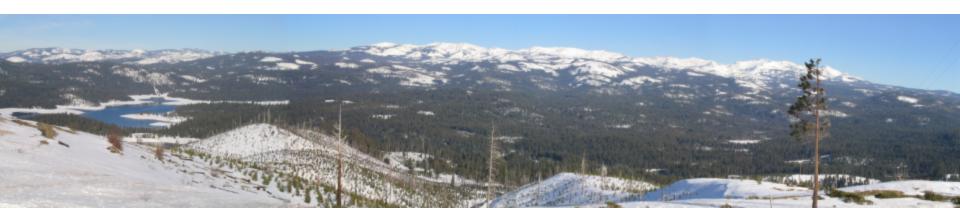
Observations of NO_2 , total peroxynitrates, total alkyl nitrates, and HNO_3 on the Western Slopes of the Sierra:

Implications for Transport of Nitrates to Lake Tahoe

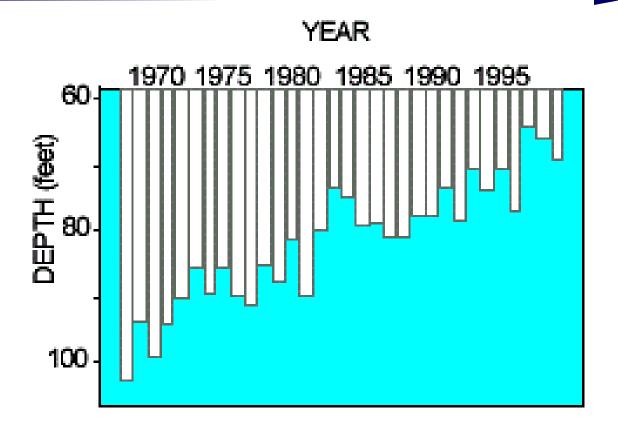


Professor Ronald C. Cohen

March 8, 2004

Department of Chemistry and Department of Earth and Planetary Science University of California, Berkeley

Visibility in Lake Tahoe



The depth at which a white disk can be seen from the surface changes from year to year but generally has become shallower over the decades.

Jassby, et al. TRG UC Davis



- Low natural nutrient concentrations
- Water has a 650 year residence time in the lake
- Visibility decline is due both to algae and mineral particulate
- Nitrogen was the limiting nutrient in 60's, now phosphorous is



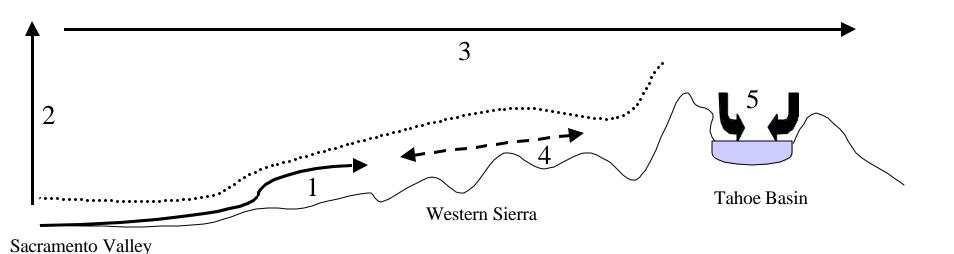
How much of the N that gets to the basin is from west of the basin?

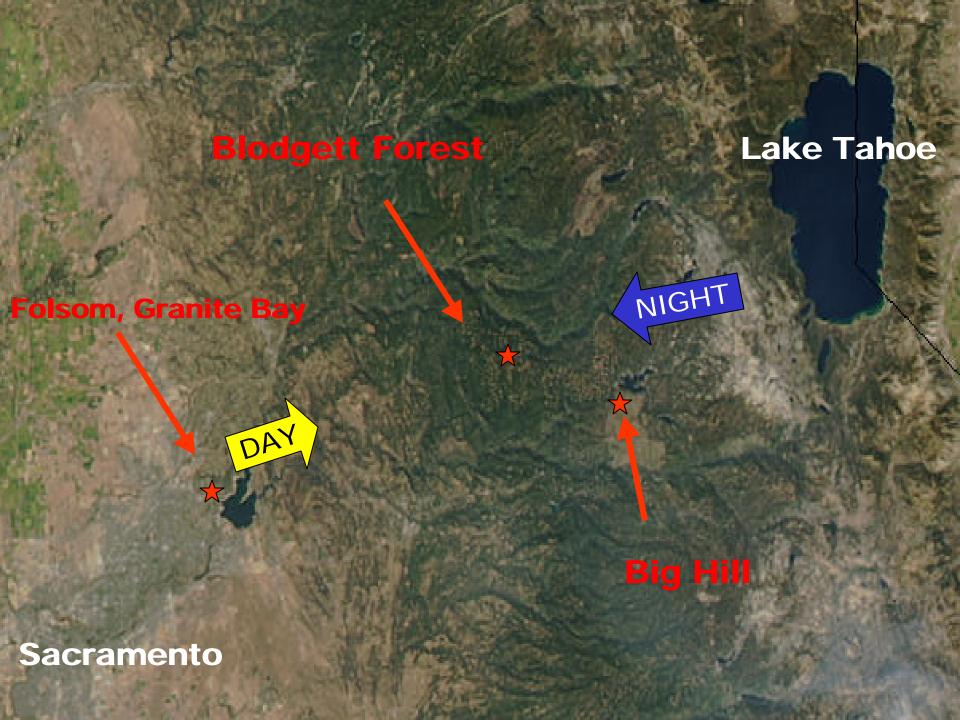
What is a general framework for understanding transport and chemistry within and surrounding the Sacramento plume?

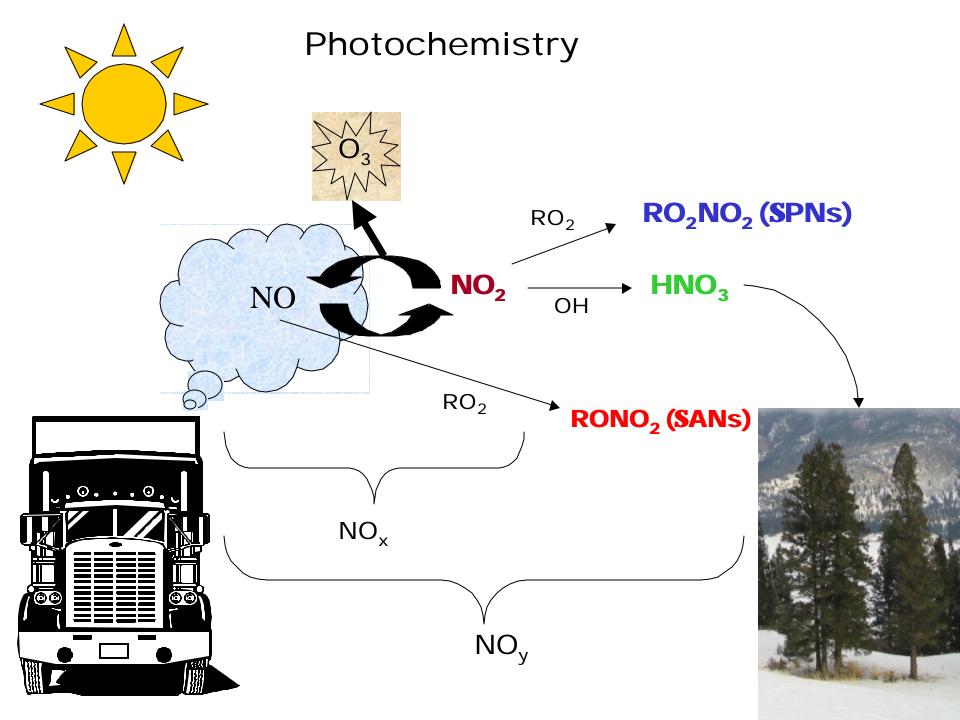
Note: Focus of this talk is on long-term mean—not on extreme events.

Possible transport paths from the West into the Tahoe basin

- Advection upslope within the boundary layer
- Vertical Mixing above the boundary layer
- Horizontal transport above the boundary layer
- Recirculation within upslope/downslope boundary layer flows
- Downward mixing within the basin





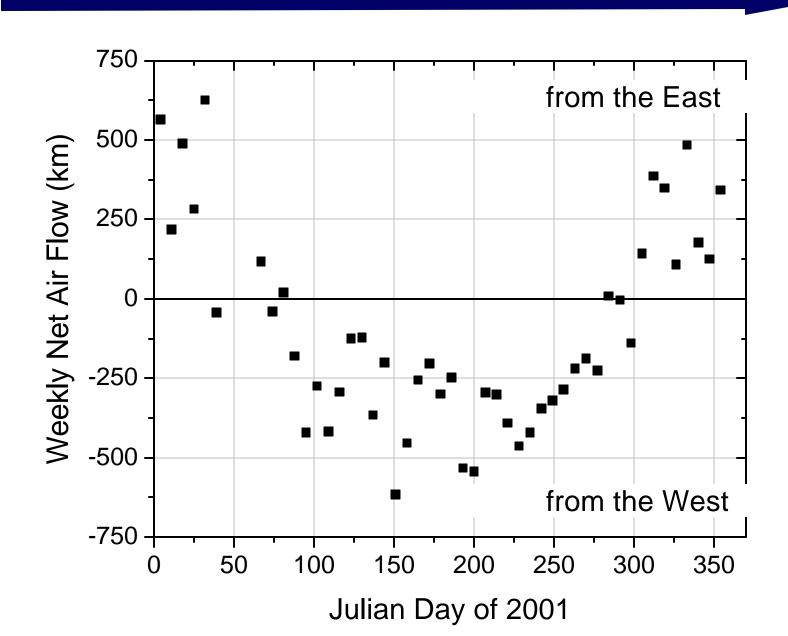


 How far does 'Sacramento plume' travel?

 How much dilution and oxidation occurs as the plume moves downwind?

- What do observations of NO_{yi} show?
- Summary and Implications

East-West Net Air Flow at UC-BFRS 2001

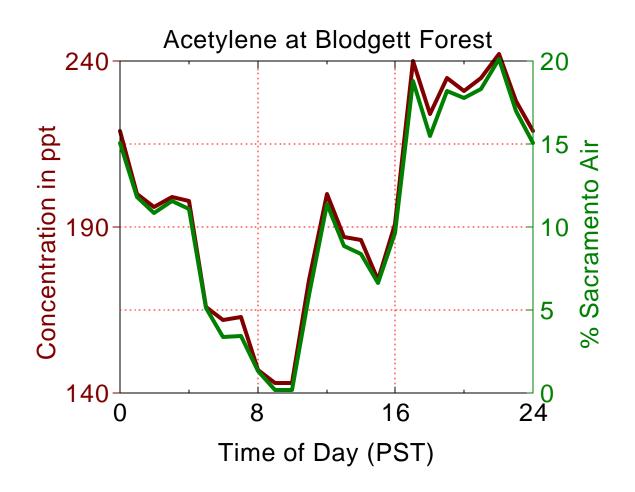


 How far does 'Sacramento plume' travel?

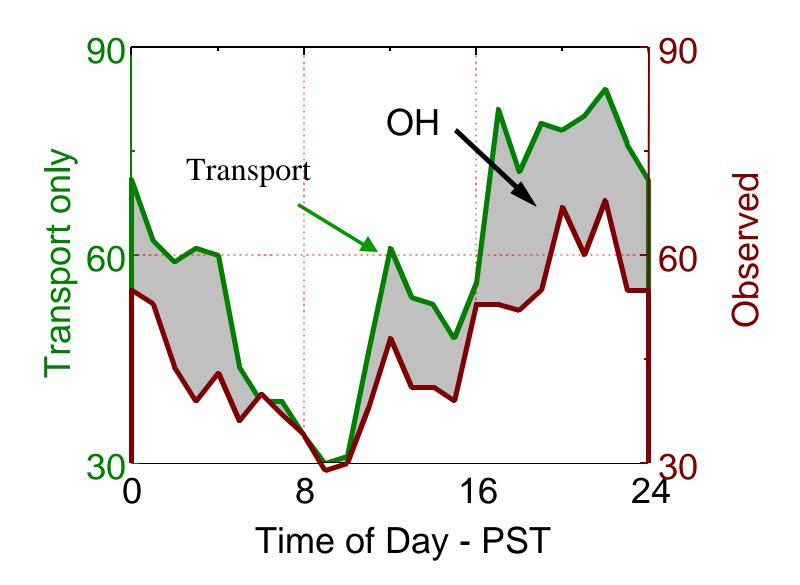
 How much dilution and oxidation occurs as the plume moves downwind?

- What do observations of NO_{yi} show?
- Summary and implications?

Long-lived hydrocarbons define the fraction of the Sacramento Plume that reaches the foothills



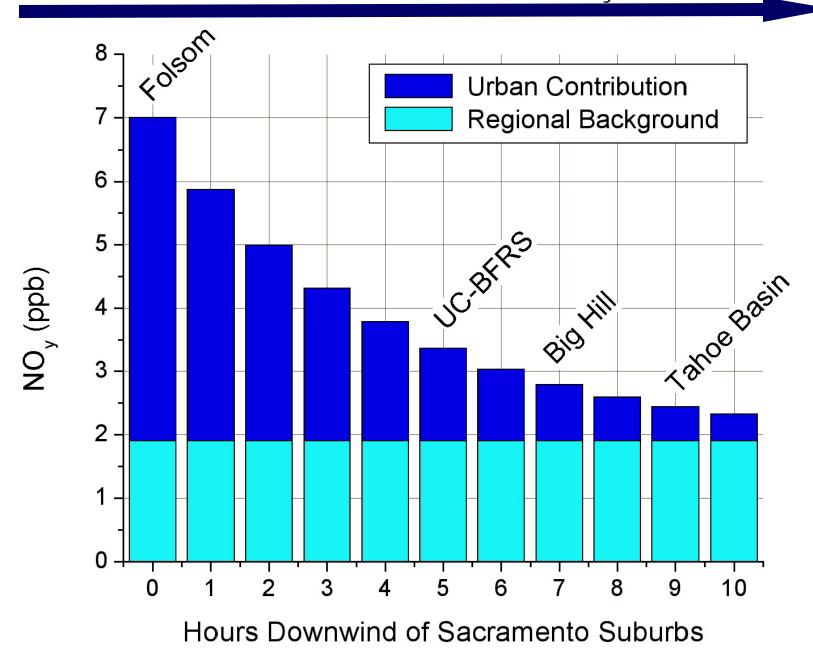
Dillon, Lammana, Schade, Goldstein and Cohen, JGR 107 2002.



Key results

- The Sacramento plume can be accurately described with a model representing advection into the foothills, dilution into the regional background and oxidation.
- The lifetime (e-folding time) with respect to dilution is 4 hours.
- An OH concentration of order 1-1.4x10⁷ molecules/cm³ describes the oxidation that occurs during plume transport.

Pure transport contribution to NO_v (an upper limit)



Implications for NO_{vi}

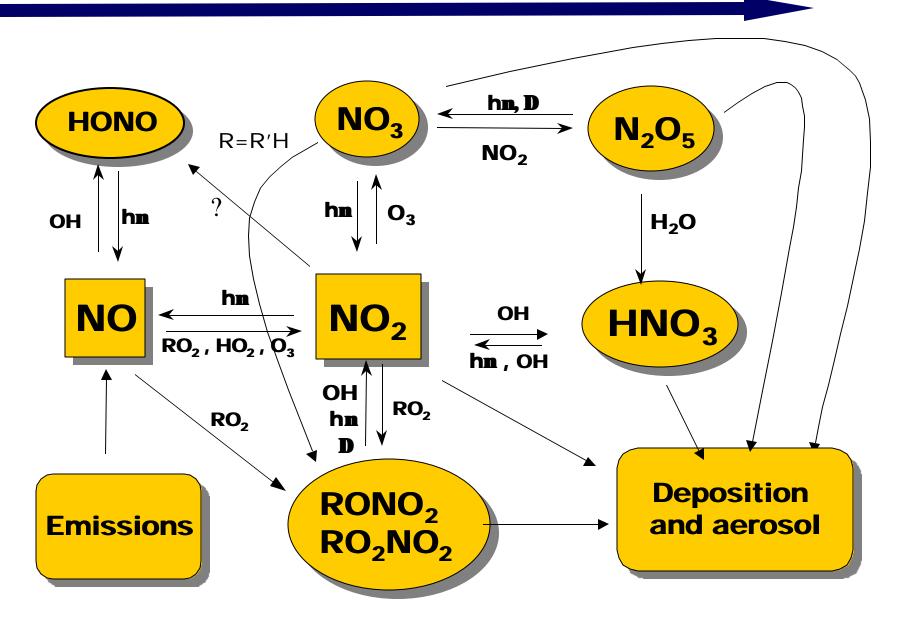
- We have a good first order model of the amount of NO_y that could be arriving from the west to the foothills of the Sierra (and perhaps as far east as Tahoe).
- The lifetime (e-folding time) with respect to dilution is 4 hours.
- Oxidation converts most of the NO₂ present in Sacramento and its suburbs to HNO₃, PANs and RONO₂ on the same time scale.

 How far does 'Sacramento plume' travel?

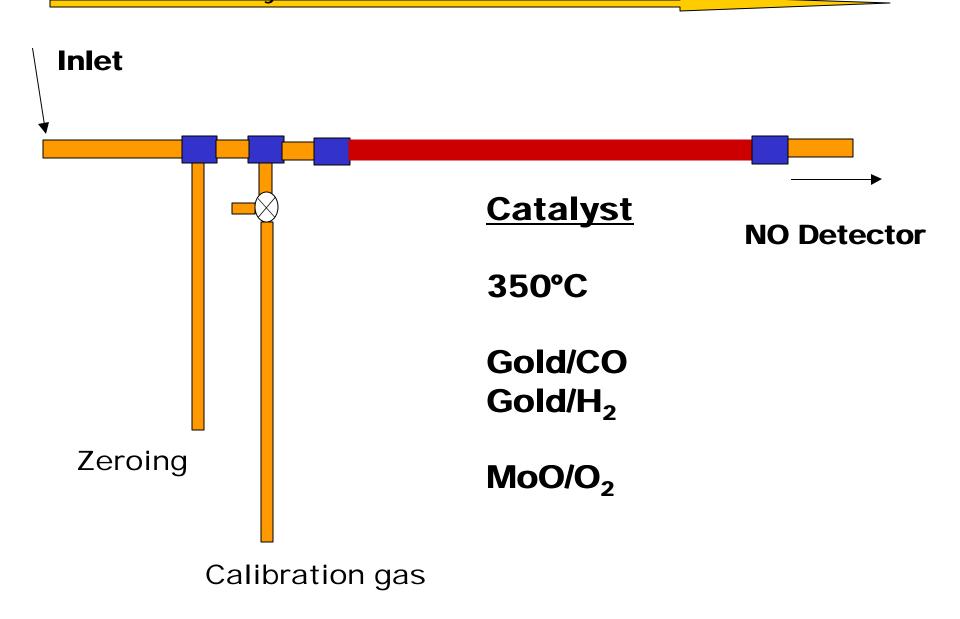
 How much dilution and oxidation occurs as the plume moves downwind?

- What do observations of NO_{yi} show?
- Summary and Implications?

NO_y



Total NO_v Measurements



NO_x measurements with catalyst

 The catalyst always converts NO₂ and some other stuff to NO.

•Typically accurate at the source and in urban areas during the night. Factors of two or more are not unusual at midday in urban areas (for example overestimates of NO_x e.g. 2ppb vs. 1ppb at noon are typical results from field comparisons).

Usually assume an overestimate of daytime
 NO_x but an underestimate of daytime NO_y.

LIF NO₂

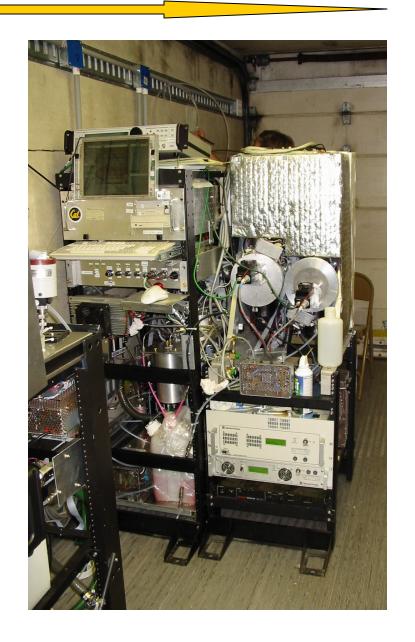
Pulsed Dye laser

Anal. Chem. 72, 528 2000

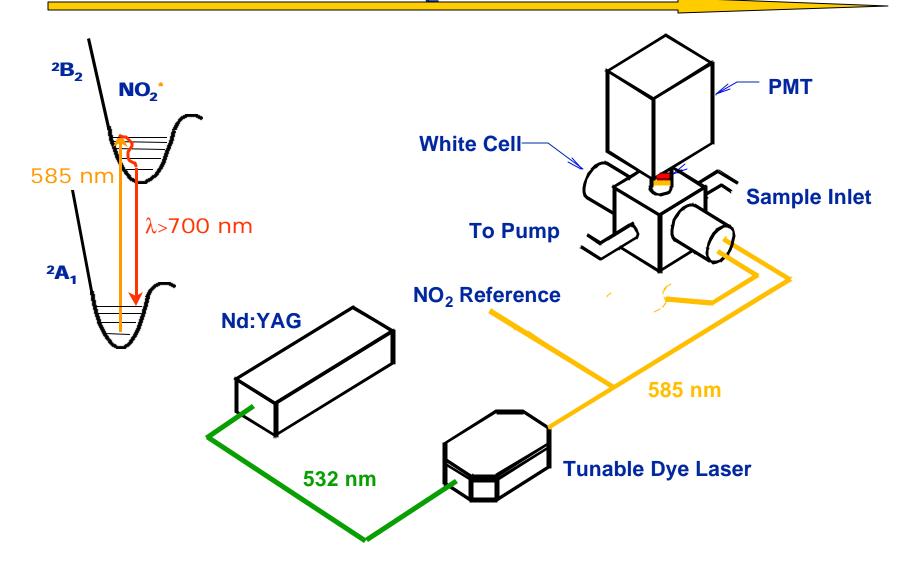
- Direct and specific
- accurate: ±5%, 1s
- sensitive: 6ppt/min, S/N=2
- detection limit 1ppt

Since 2000

- sensitivity <1ppt/min
- ~7 days unattended

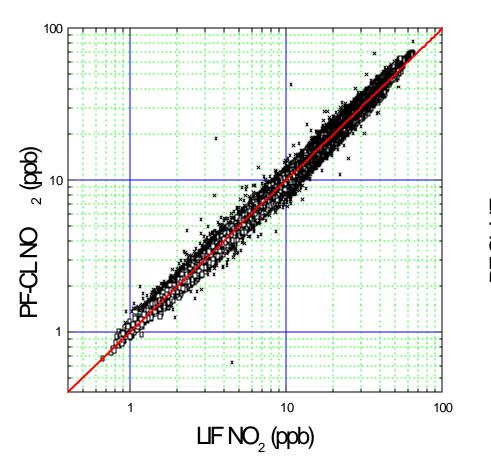


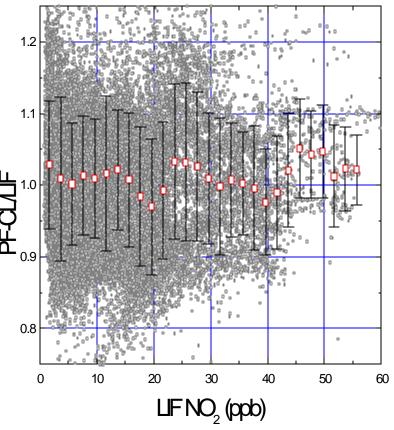
LIF detection of NO₂



Analytical Chemistry, 72, 528, 2000

NO₂ Instrument Comparison (SOS 99)







- 638nm laser and a ss-jet
- sensitivity of 75ppt/min
- 130lbs
- weeks without an operator

Diode laser (Patti Cleary)

Applied Optics **41**(33), 6950, 2002

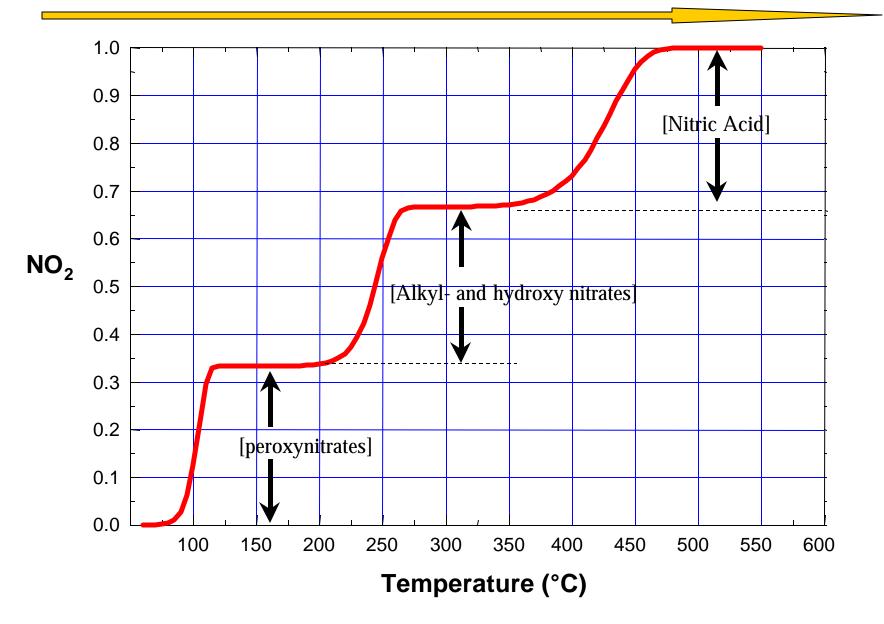


ΣPNs, ΣANs, & HNO₃ + heat \rightarrow RO_x + NO₂

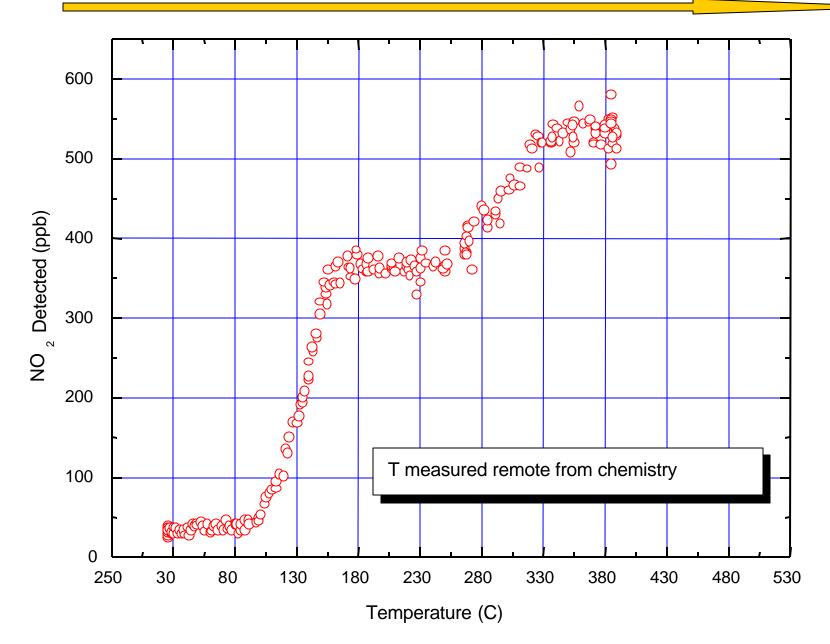
Thermal
Dissociation-LIF
Doug A. Day, et al,
JGR., 107(D6), '02.

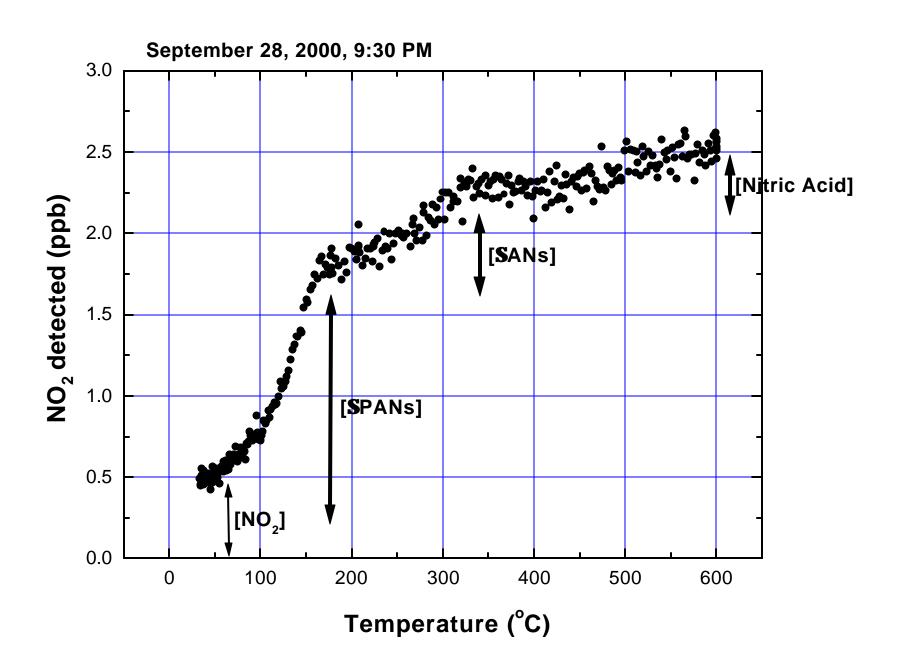


ΣPNs, ΣANs, & HNO₃ + heat \rightarrow RO_x + NO₂

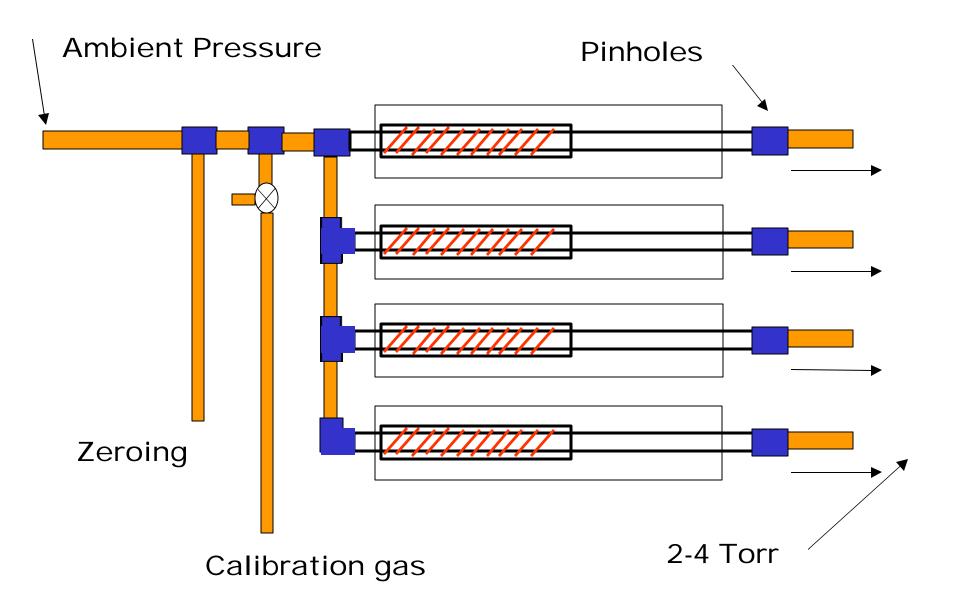


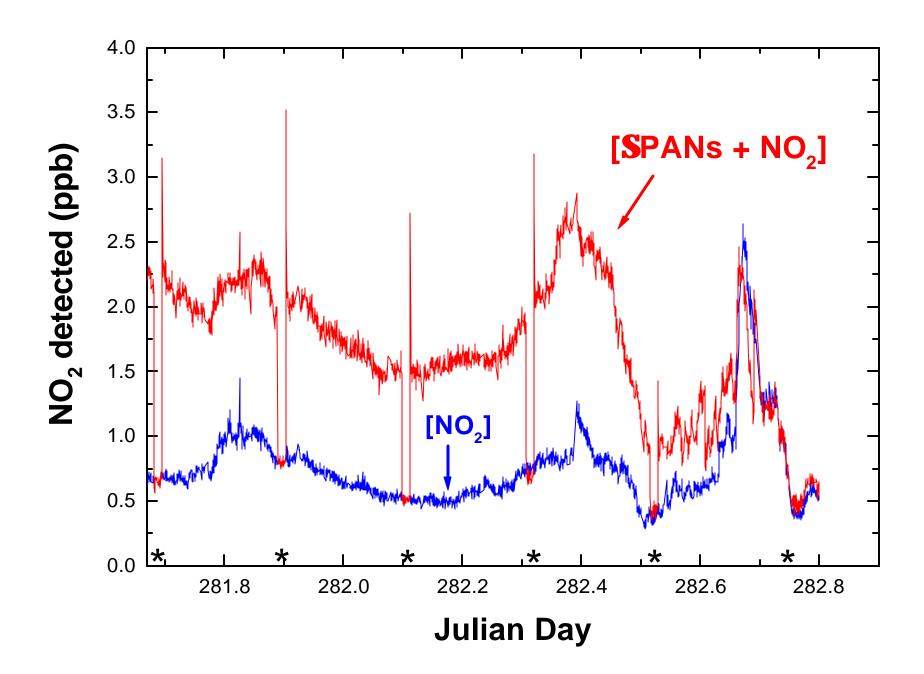
HONO₂ and C₂H₅ONO₂





Inlet and Dissociation Heaters





TD-LIF

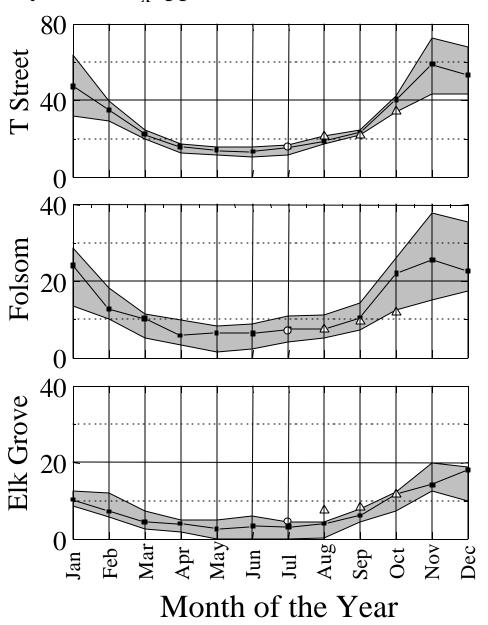
Accuracy 15%

Sensitivity ~10 ppt

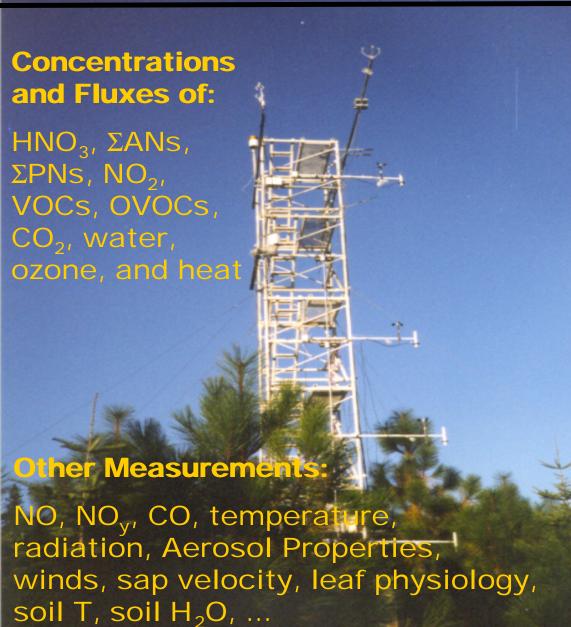
Used at UC-Blodgett Forest, CA Granite Bay, CA Big Hill, CA Houston, TX and from the NCAR C-130 (0-7km ASL 35°-85°N Latitude)



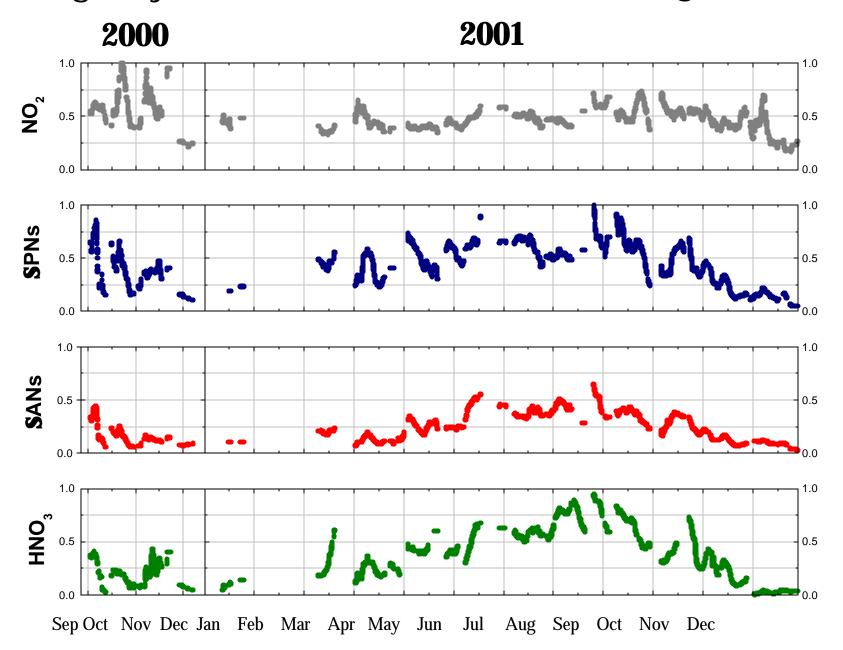
Daytime NO_x (ppb) in the Sacramento Area



Blodgett Forest 12 m Tower



Blodgett Forest



Big Hill Observations from Tower atop trailer

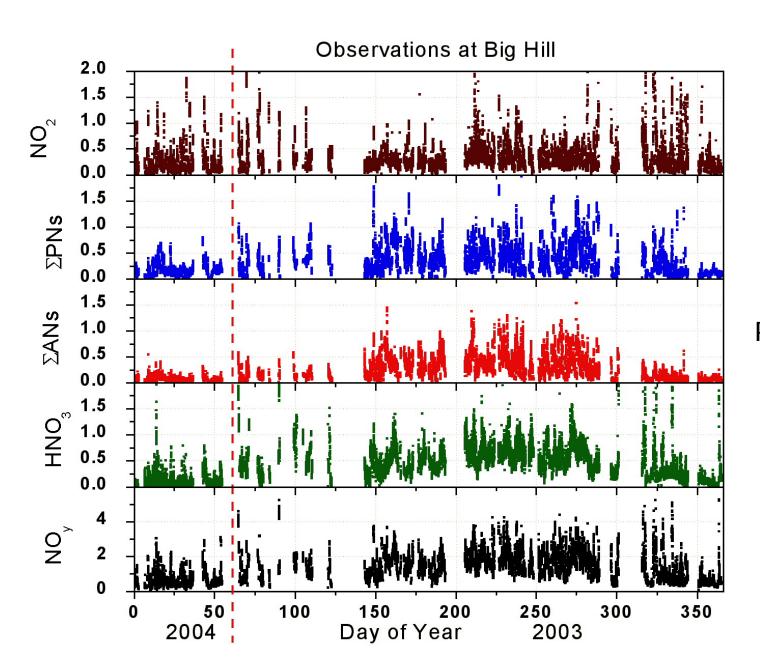








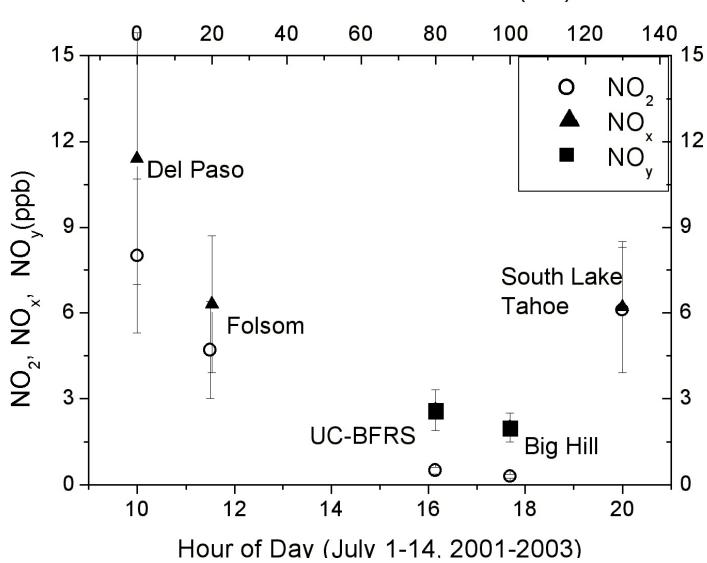
Jennifer Murphy and Paul Wooldridge



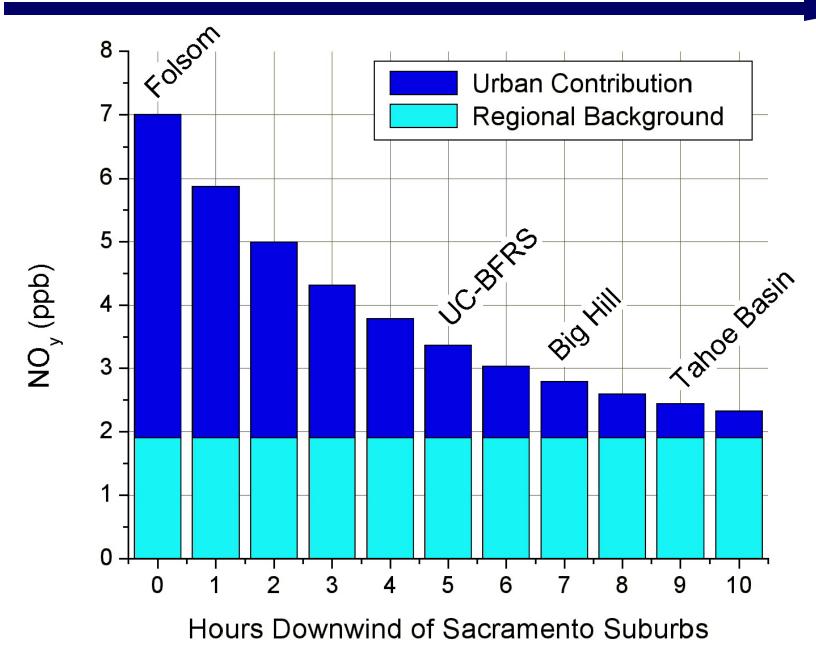
Preliminary

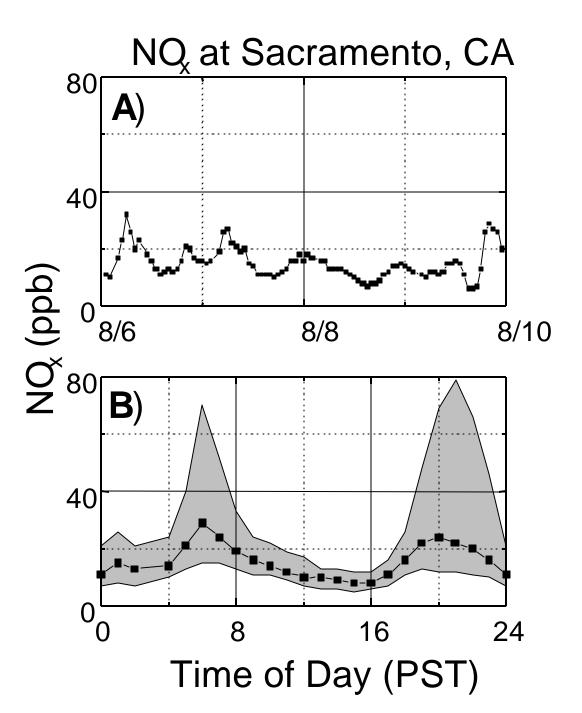
Summertime plume

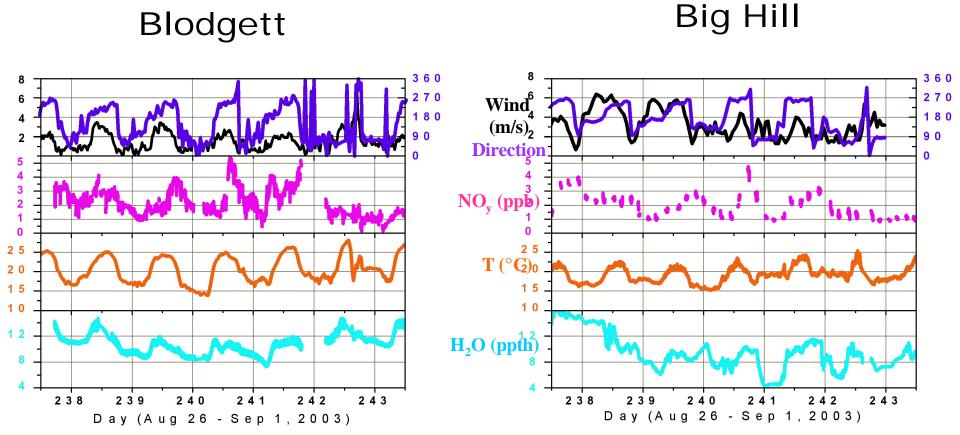




Pure transport contribution to NO_y (an upper limit)



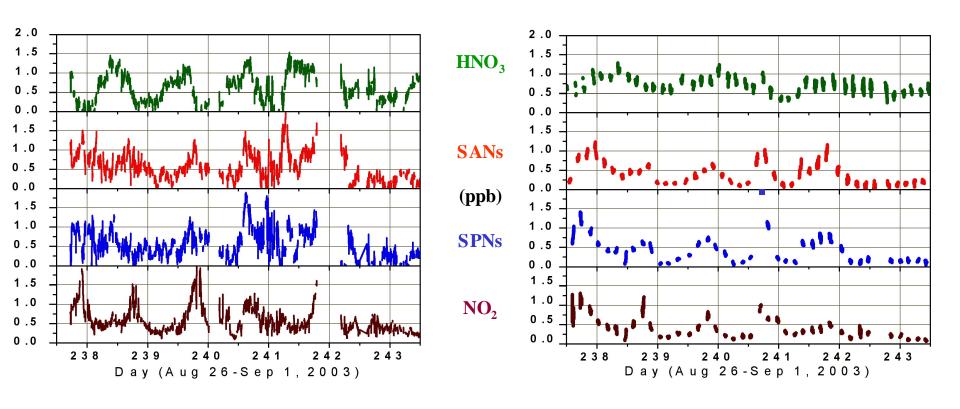




Blodgett has bigger T swings, smaller H₂O swings, slower winds.

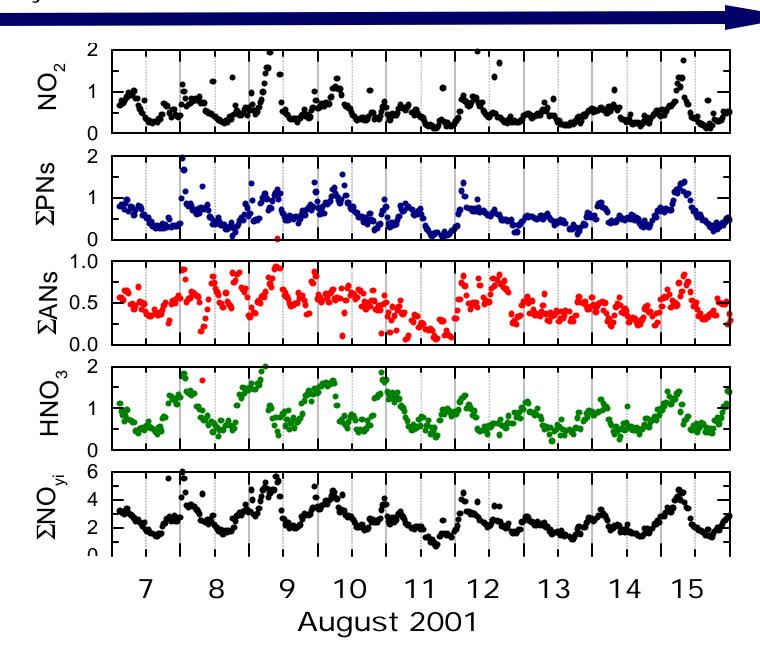


Big Hill

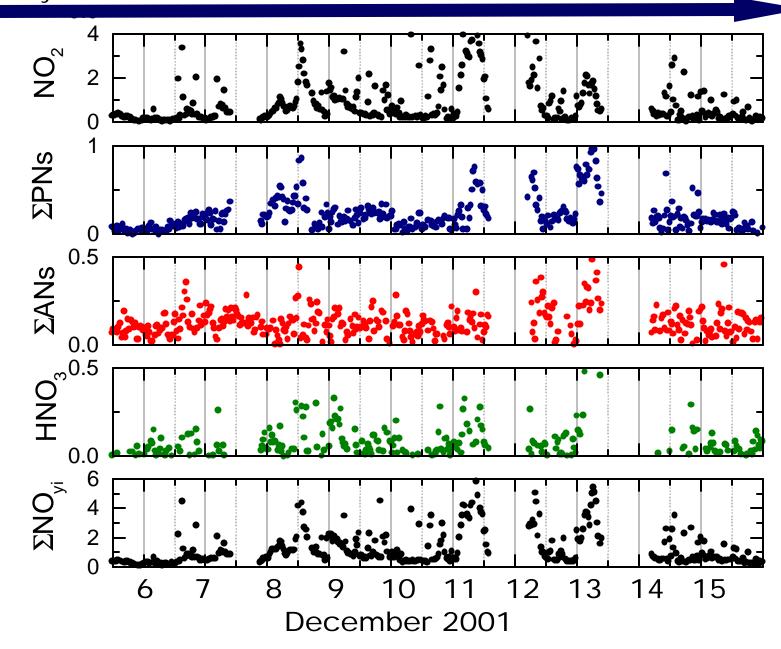


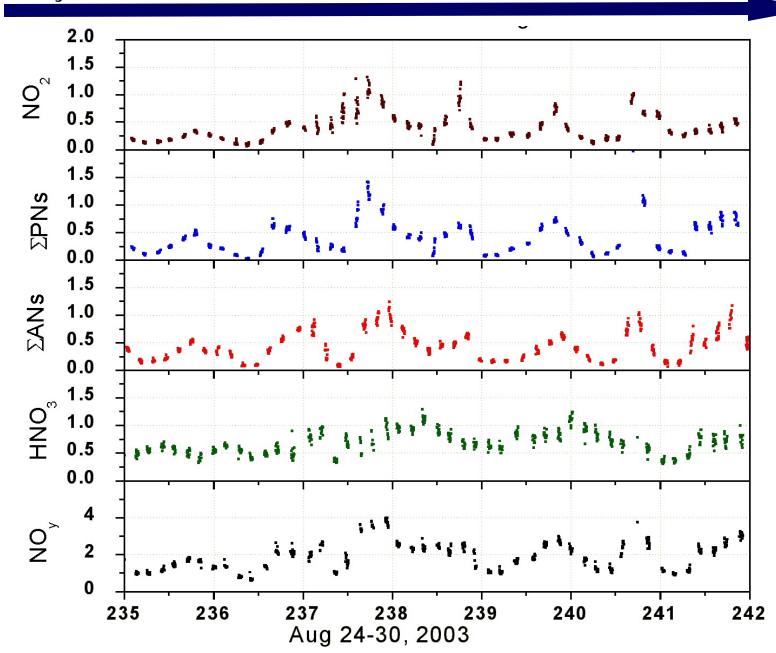
More NO₂ at Blodgett suggests less processed, younger air. HNO₃ at Big Hill doesn't disappear at night.

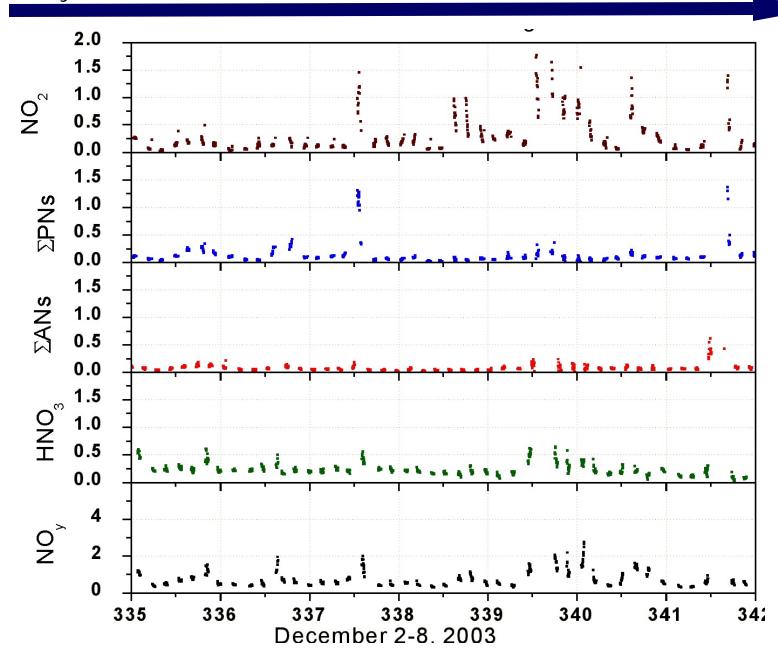
NO_{yi} at UC Blodgett Forest



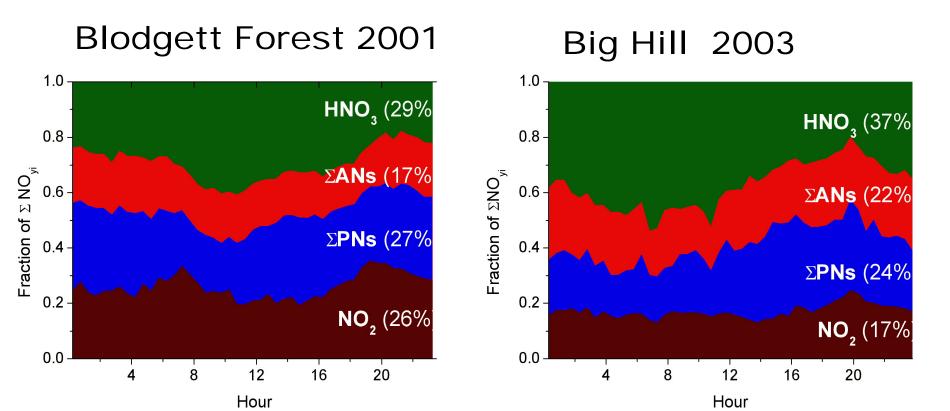
NO_{yi} at UC Blodgett Forest





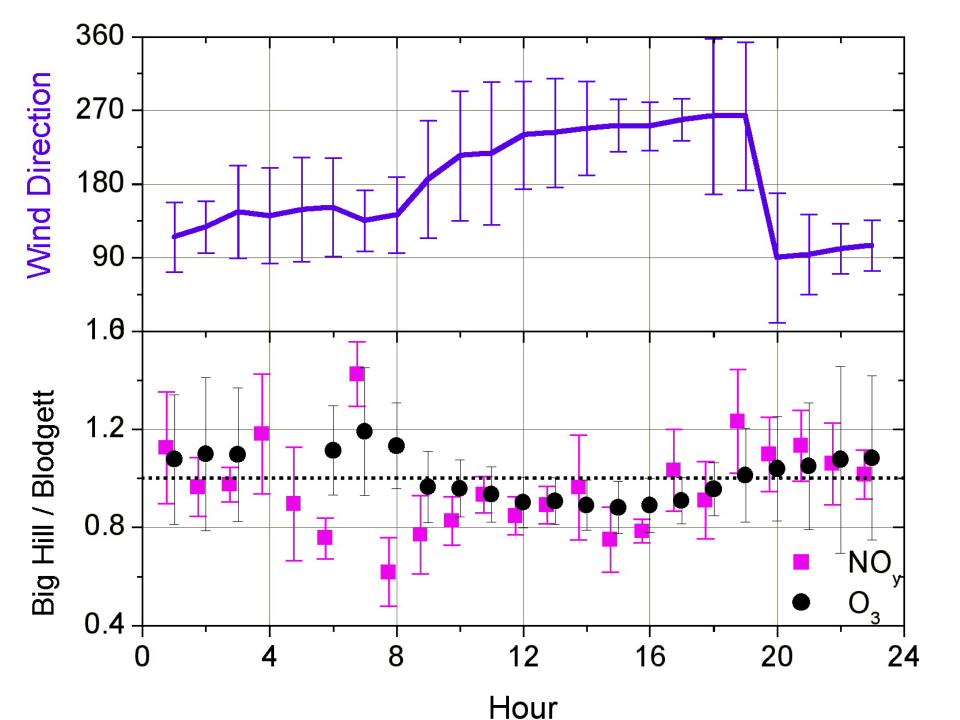


Summer



The two sites sample similar air during the day.

Reactive nitrogen at Big Hill is more oxidized and diluted.



HNO₃

Day

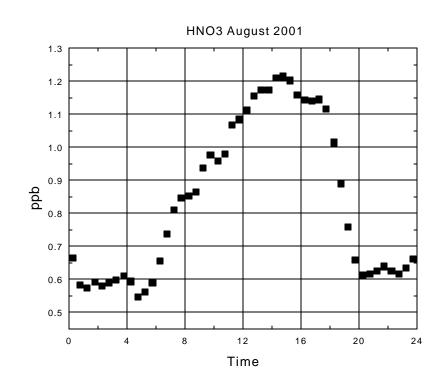
$$k_{OH+NO2}$$
 [OH][NO₂] $\cong k_{deposition}$ [HNO₃] or k_{OH+NO2} [OH]/ $k_{deposition} \cong$ [HNO₃]/[NO₂]

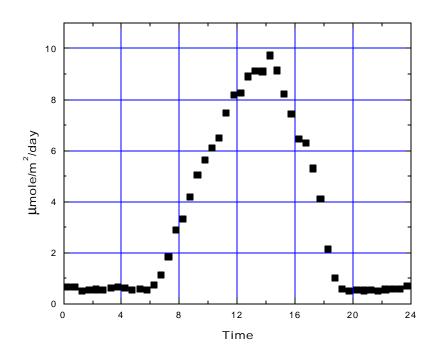
Expect [HNO₃]/[NO₂] to follow OH since k's are roughly constant and fast.

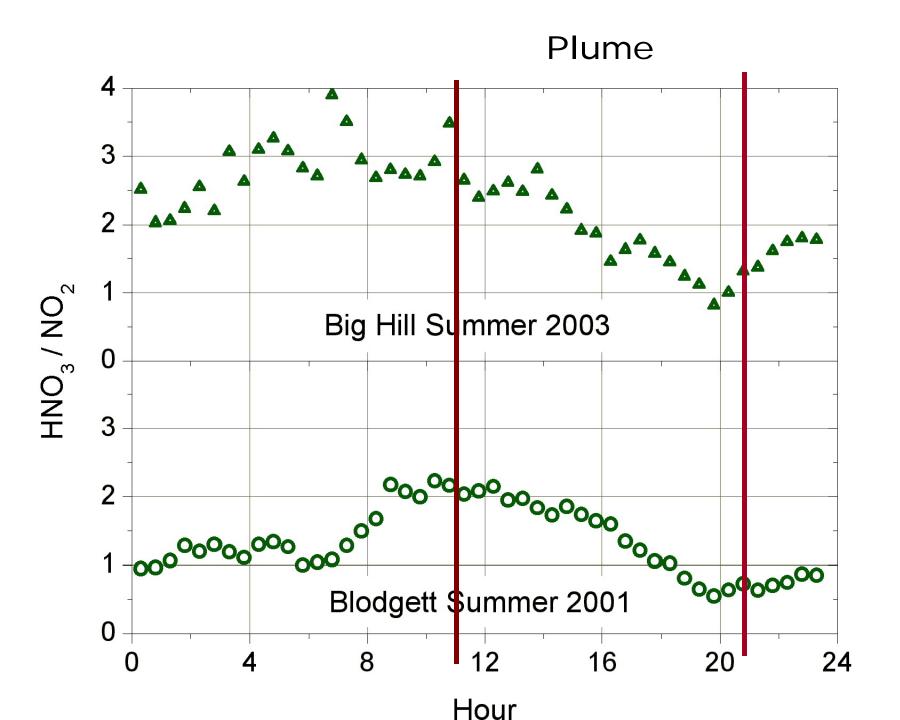
Night

Chemistry of NO₃ and deposition approaches zero.

Blodgett Forest deposition of HNO₃







HNO₃/NO₂ slightly higher at Big Hill than Blodgett, implying deposition is less efficient or OH concentration is higher.

The two sites cannot be perfectly modeled using a single transport path with one effective deposition velocity.

Σ ANs and Σ PNs

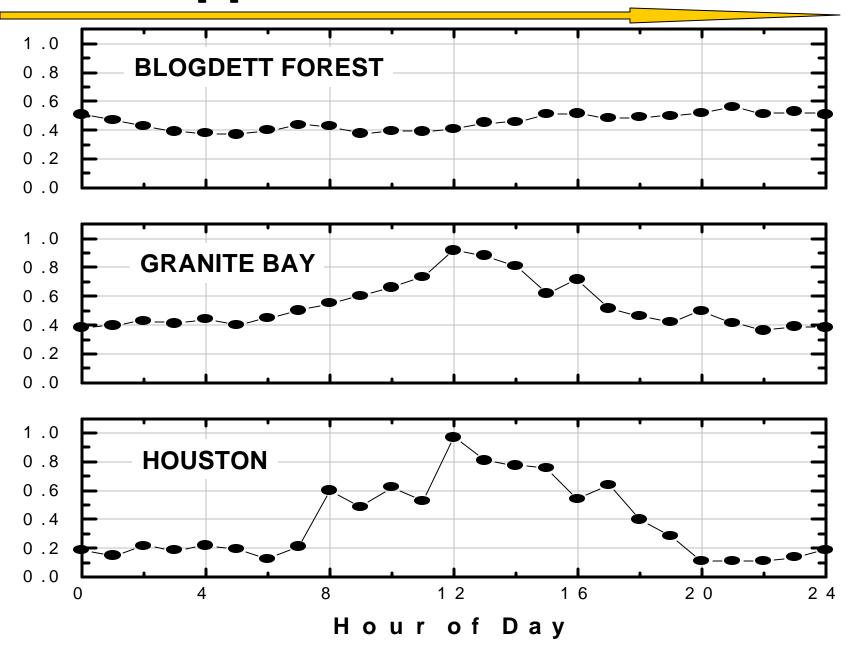
Day

Measurements show concentrations follow OH in urban source regions but are decoupled downwind. Implies deposition of Σ ANs is rapid (<4 hours) and thermal decomposaition of Σ PNs is rapid (<4 hours).

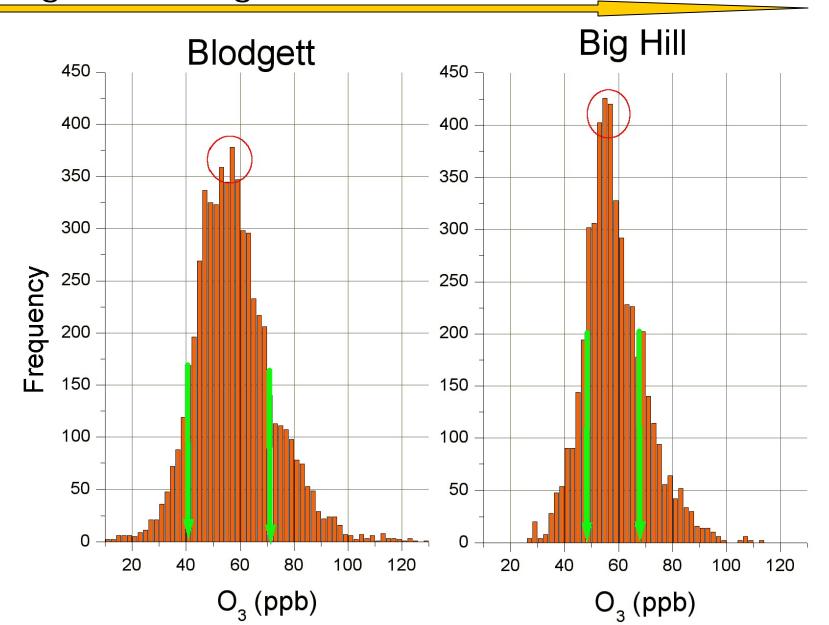
Night

Chemistry of NO₃ and deposition approaches zero.

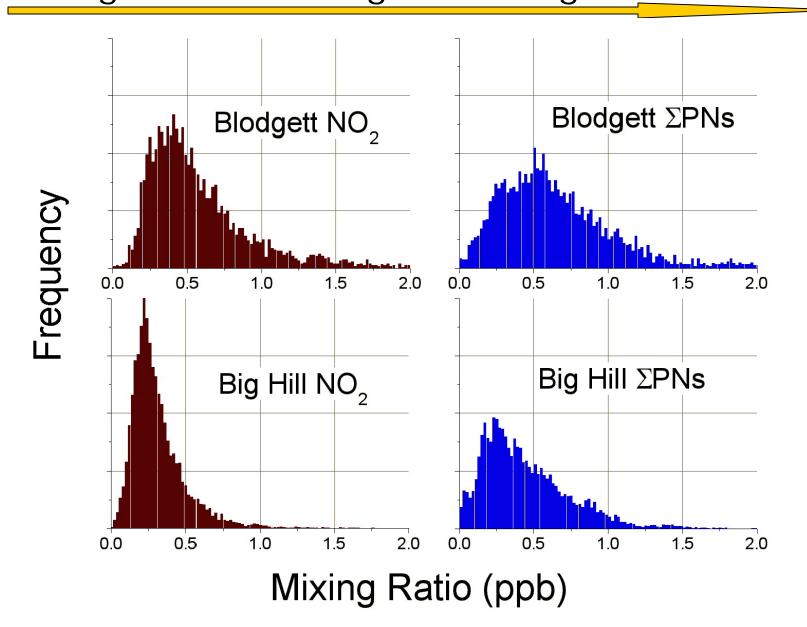
SANs ppb



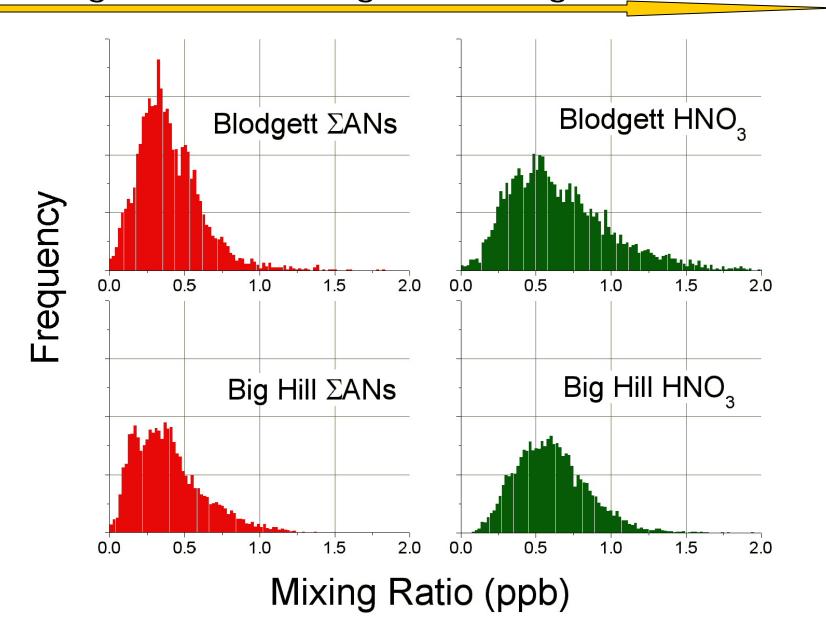
The observed variance of O₃ mixing ratios at Blodgett and Big Hill



The observed variance of NO_2 and ΣPN mixing ratios at Blodgett and Big Hill

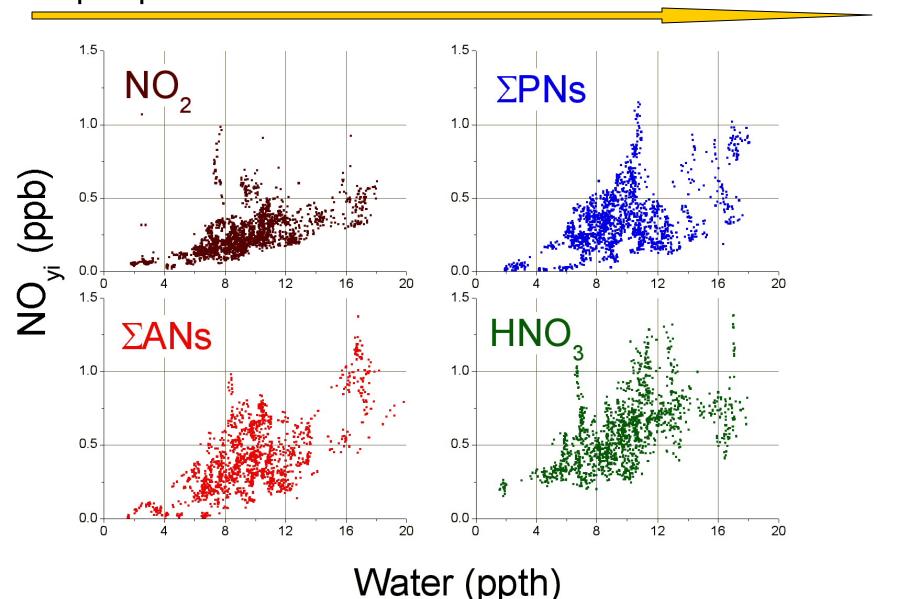


The observed variance of NO_2 and ΣPN mixing ratios at Blodgett and Big Hill



 The variance of the NO_{yi} chemicals is proportional to the local NO₂ concentration.

 Extreme events are even less important at Big Hill than at Blodgett. Correlations of water and NO_y at Big Hill (July '03) suggest mixing of air from the valley with the free troposphere



It may be possible to use these correlations and observations of water mixing ratio at other sites down wind of Big Hill to estimate potential transport fluxes.

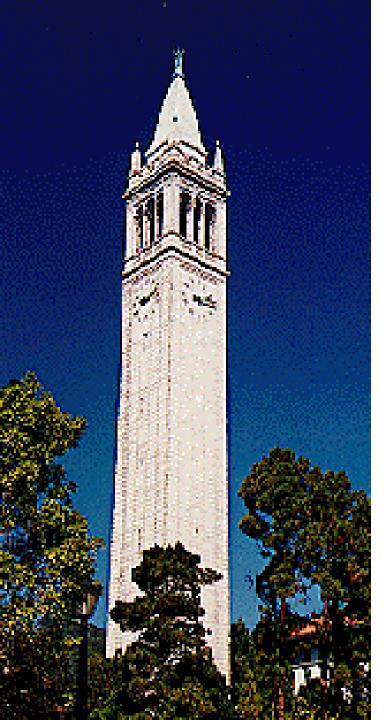
- How far does 'Sacramento plume' travel in one day?
- How much dilution and oxidation occurs as the plume moves downwind?

- What do observations of NO_{yi} show?
- Summary and Implications?

Conclusions

- Transport from the west is an extremely small contribution to the NO_y in the air at Lake Tahoe on the daily time scale—factors that contribute to a large regional background that accumulates over multiple days of NO_y may contribute. The source of this background is not understood.
- During the day HNO₃ has a lifetime of about 3 hours and its mixing ratio is controlled by OH, NO₂ and deposition rates. Since OH is strongly buffered, local (within 3 hrs transport time) sources of NO_x and are the major atmospheric variable affecting HNO₃ dry deposition fluxes.
- We don't yet have a complete story explaining ΣPNs and ΣANs, but to the extent that they do deposit rapidly they are unlikely to be a source for N-transported into the basin.
- There is a wealth of high time resolution annual and multiyear data to assess chemistry and transport of nitrogen oxides along the western slopes of the Sierra.





Cohen Group

grad students

Ezra Wood
Rebecca Rosen
Jennifer Murphy
Delphine Farmer
Chris Cappa
Tim Bertram
Chika Minejima
Idalia Perez
Anne Perring

Doug Day PhD '03 Patti Cleary PhD '03 Michael Dillon PhD '02 Joel Thornton PhD '02 Melissa Hendricks PhD '00

Staff
Paul Wooldridge